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Preliminary communication

## Studies on organolanthanide complexes

## XXXIX<sup>†</sup>. Synthesis and molecular structure of $(MeCp)_{3}La$ (MeCp = CH<sub>3</sub>C<sub>5</sub>H<sub>4</sub>): a tetrameric complex of the type [(MeCp)<sub>3</sub>La]<sub>4</sub>

Zuowei Xie, F. Ekkehardt Hahn<sup>1</sup> and Changtao Qian \*

Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Lu, Shanghai 200032 (People's Republic of China)

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## Abstract

Reaction of LaCl<sub>3</sub> with excess Na(MeCp) in THF, followed by sublimation, affords (MeCp)<sub>3</sub>La. The complex crystallizes as a tetramer [(MeCp)<sub>3</sub>La]<sub>4</sub> in the monoclinic space group  $P2_1/c$  with unit cell dimensions a 9.6785(14), b 26.058(5), c 12.506(2) Å,  $\beta$  97.724(13)° and Z = 2 (tetramers).

Up to now, the molecular structures of the following tricyclopentadienyl lanthanide complexes have been reported: Cp<sub>3</sub>Sc [2], Cp<sub>3</sub>Pr [3], Cp<sub>3</sub>Nd [4], Cp<sub>3</sub>La [5], Cp<sub>3</sub>Er [6], Cp<sub>3</sub>Tm [6], Cp<sub>3</sub>Yb [7], Cp<sub>3</sub>Lu [8], (MeCp)<sub>3</sub>Nd [9], (MeCp)<sub>3</sub>Yb [10], (MeCp)<sub>3</sub>Ce [11], and [(Me<sub>3</sub>Si)<sub>2</sub>C<sub>5</sub>H<sub>3</sub>]<sub>3</sub>Ce [11]. The results of above structural analyses suggest that the solid-state structures of the trivalent lanthanide metallocenes are dependent on the lanthanide metals as well as the size of the substituents on the cyclopentadienyl rings. For example, (MeCp)<sub>3</sub>Nd is tetrameric while Cp<sub>3</sub>Nd is a polymer [9,10]; (MeCp)<sub>3</sub>Ce is tetrameric while [(Me<sub>3</sub>Si)<sub>2</sub>C<sub>5</sub>H<sub>3</sub>]<sub>3</sub>Ce and (Me<sub>3</sub>SiC<sub>5</sub>H<sub>4</sub>)<sub>3</sub>Ce are monomers [11]. An interesting question is how is the molecular structure of (MeCp)<sub>3</sub>La compared to the polymeric  $\eta^2 : \eta^5$ -Cp<sub>3</sub>La [5].

We report the crystal structure of solvent-free  $(MeCp)_3La$  (1) prepared by the reaction of LaCl<sub>3</sub> with excess (MeCp)Na in THF, followed by vacuum sublimation [12\*]. Single crystals suitable for an X-ray study were obtained from a toluene solution at room temperature.

<sup>&</sup>lt;sup>†</sup> For Part XXXVIII see ref. 1.

<sup>&</sup>lt;sup>1</sup> Visiting scholar from Department of Chemistry, Technische Universität Berlin, Straße des 17. Juni 135, W-1000 Berlin 12, Germany. Address inquiries regarding X-ray diffraction results to this author.

<sup>\*</sup> Reference number with asterisk indicates a note in the list of references.



Fig. 1. ORTEP drawing of one tetramer  $[(MeCp)_3La]_4$  (two asymmetric units). Starred atoms represent transformed coordinates of the type 1-x, -y, -z. Cp denotes the centroids of the cyclopentadienyl rings. Important bond distances (Å) and angles (°): LaA-Cp1A 2.553(7), LaA-Cp2A 2.569(7), LaA-Cp3A 2.637(7), LaB-Cp1B 2.601(7), LaB-Cp2B 2.555(7), LaB-Cp3B 2.630(7), LaA-C15B 2.972(5), LaB-C15A\* 3.064(6), Cp1A-LaA-Cp2A 116.5(2), Cp1A-LaA-Cp3A 121.4(2), Cp2A-LaA-Cp3A 113.8(2), Cp1B-LaB-Cp2B 119.5(2), Cp1B-LaB-Cp3B 114.5(2), Cp2B-LaB-Cp3B 115.0(2), LaA-C15A-LaB\* 168.8(3), LaB-C15B-LaA 178.5(2).

The X-ray structure analysis [13\*] shows that  $(MeCp)_3La$  crystallizes as a tetramer,  $[(MeCp)_3La]_4$ . Figure 1 shows one tetramer with the adopted numbering scheme. The structure of 1 differs significantly from that of Cp<sub>3</sub>La [5]. The methyl substituent on the cyclopentadienyl ring forces the mode of coordination from an  $\eta^2$ :  $\eta^5$  polymer for Cp<sub>3</sub>La [5] to an  $\eta^1$ :  $\eta^5$  tetramer for 1.

In  $(MeCp)_3La$ , each La atom is surrounded by two terminal  $\eta^5$ -MeCp groups and a bridging MeCp group which is  $\eta^5$ -bonded to one lanthanum atom and  $\eta^1$ -bonded to an adjacent lanthanum atom. Thus the coordination number of the La atoms is ten. The geometry of tetrameric  $(MeCp)_3La$  is nearly identical with those of its cerium and neodymium analogues  $(MeCp)_3Ln$  (Ln = Ce (2), Nd(3)) [9,11]. The average  $\eta^1$ -C-Ln distance is 3.018(7) Å in 1, 3.03(3) Å in 2 and 2.984(3) Å in 3. The Ln-C( $\eta^5$ -MeCp) distance in 1 is 2.843(3) Å, in 2 it is 2.83(4) Å and in 3 it is 2.79(4) Å. The Cp-Ln-Cp angles (Cp denotes the centroids of the cyclopentadienyl rings) are almost identical for all three derivatives at 117°. The differences in bond length are most reasonably ascribed to the lanthanide contraction.

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## **References and notes**

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- 12 Reaction of 3.17 g (12.92 mmol) of LaCl<sub>3</sub> with 33.0 ml (50.10 mmol) of Na(MeCp) in 70 ml of THF overnight, followed by sublimation at 200-230 °C/10<sup>-3</sup> mmHg, gave white crystals, 2.50 g (yield 51.5%). Anal. Found: C, 57.22; H, 5.84. C<sub>18</sub>H<sub>21</sub>La calc.: C, 57.46; H, 5.63%. MS (EI): 751 (2*M*<sup>+</sup> 1), 673 ([2*M* MeCp]<sup>+</sup>), 376 (*M*<sup>+</sup>), 297 ([(MeCp)<sub>2</sub>La]<sup>+</sup>), 217 ([(MeCp)La]<sup>+</sup>).
- 13 Crystal data for [(MeCp)<sub>3</sub>La]<sub>4</sub>:  $C_{72}H_{84}La_4$ , M = 1505.12. Crystals are monoclinic, space group  $P2_1/c$  (no. 14). a 9.6785(14), b 26.058(5), c 12.506(2) Å.  $\beta 97.724(13)^\circ$ . V 3125(2) Å<sup>3</sup>.  $d_c 1.60$  g cm<sup>-3</sup> for Z = 2,  $\mu 27.3$  cm<sup>-1</sup>. Data collection on a CAD-4 diffractometer at 24(4)°C with Mo- $K_a$  radiation ( $\lambda 0.71073$  Å), crystal size  $0.35 \times 0.38 \times 0.42$  mm,  $2\theta$  range  $2^\circ \le 2\theta \le 50^\circ$ , systematic absences uniquely defined the space group to be  $P2_1/c$ . Data reduction with Lorentz, polarization and absorption correction gave 5031 unique reflections. Solution with Patterson methods, refinement with Fourier and least-squares methods gave R = 0.032,  $R_w = 0.053$  and GOF = 2.084 for 4339 reflections  $F_0^{-2} \ge 3\sigma(F_0^{-2})$  and 343 parameters.